

Equilibrium combustion products: Dissociation effects

P=30 atmospheres

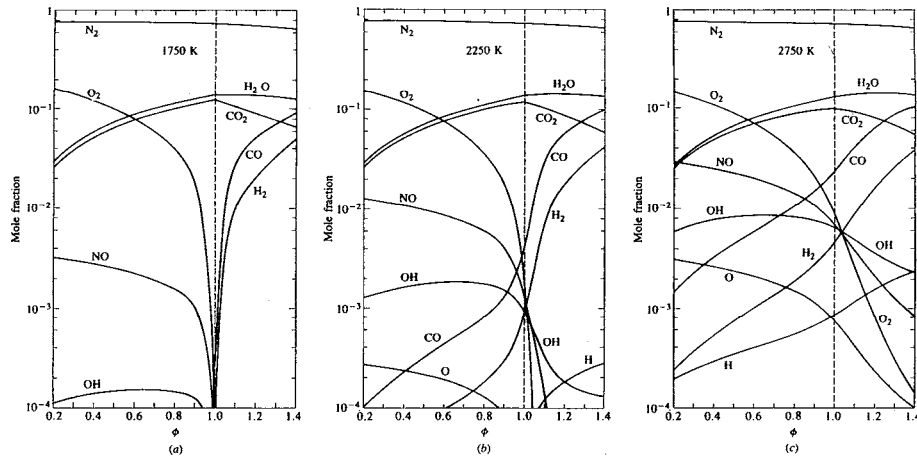


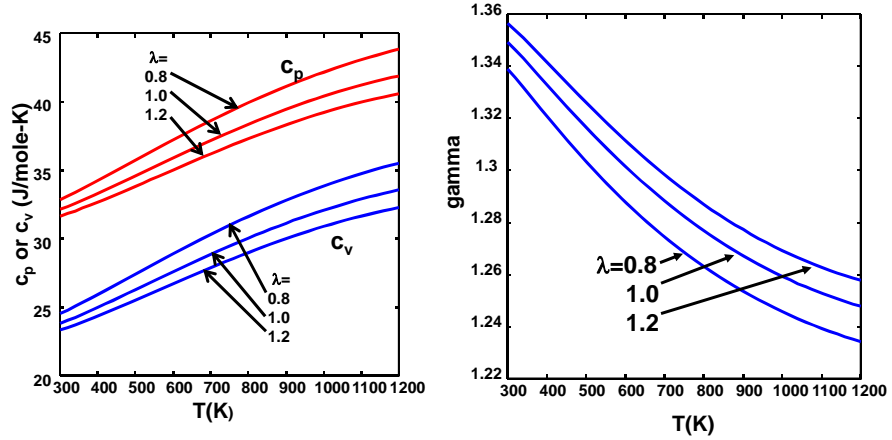
FIGURE 3-10
Mole fractions of equilibrium combustion products of isoctane-air mixtures as a function of fuel/air equivalence ratio at 30 atmospheres and (a) 1750 K; (b) 2250 K; and (c) 2750 K.

© McGraw-Hill Education. All rights reserved. This content is excluded from our Creative Commons license. For more information, see <https://ocw.mit.edu/help/faq-fair-use>.

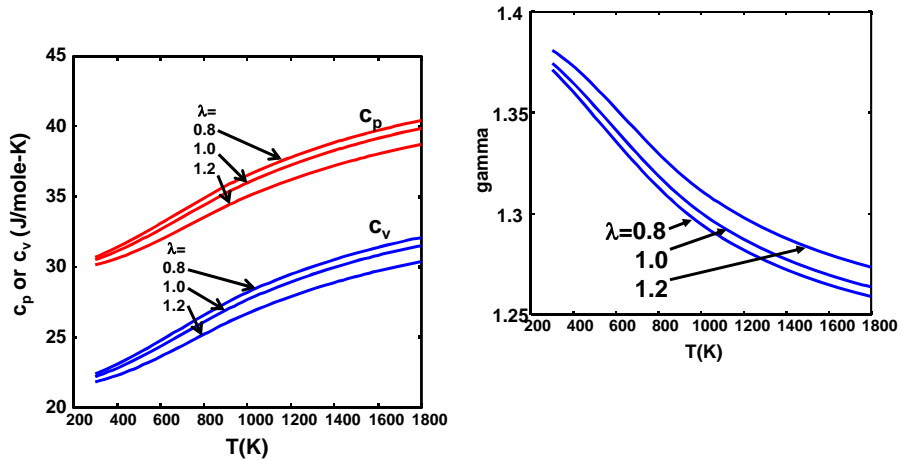
Thermodynamic model of engine charge for heat release process

- Unburned gas
 - Ideal gas of frozen composition
- Burned gas
 - At high temperature ($T > 1740\text{K}$), as equilibrium mixture
 - At low temperature ($T < 1740\text{K}$), as frozen mixture

Unburned gas properties for gasoline (CH_{1.85})/air



Burned gas properties for gasoline (CH_{1.85})/air



Composition frozen at 1740K

Fuel-air cycle results

In the Fuel-Air Cycle, the engine processes are still modeled as ideal but the properties of the working fluid (fuel/air/residual gas mixture before combustion, and burned gases in chemical equilibrium after combustion) are described accurately.

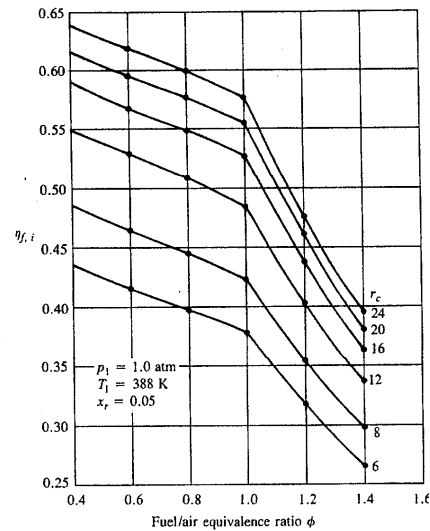
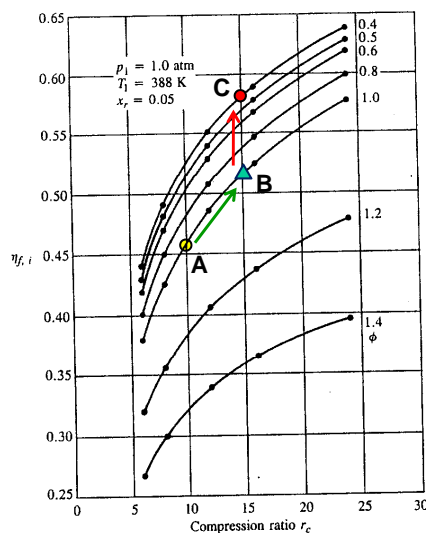
The results from this improved cycle analysis model are useful for estimating, approximately, the effects of compression ratio, fuel/air equivalence ratio, and mixture inlet conditions on engine efficiency and performance. The following approximate relationships are useful.

1. The maximum indicated fuel conversion efficiency of an actual engine is about 0.85 times the efficiency of the equivalent fuel-air cycle.
2. Results from change of engine operating condition can be interpreted in terms of percentage change in output values

Computer codes which accurately simulate the real engine cycle have now been developed and are widely used.

Fuel-air cycle results: $\eta_{f,i}$

Fuel: octene; $p_1 = 1 \text{ atm}$, $T_1 = 388 \text{ K}$, $x_r = 0.05$ (Fig. 5.9)

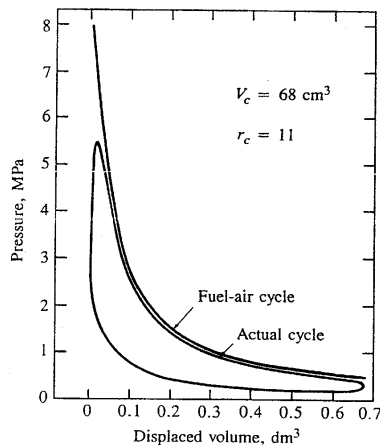


A=SI engine at stoichiometric with $r_c=10$; C=Diesel at A/F=36 ($\Phi=0.4$) with $r_c=15$

© McGraw-Hill Education. All rights reserved. This content is excluded from our Creative Commons license. For more information, see <https://ocw.mit.edu/help/faq-fair-use>.

Real Cycle Effects

1. **Combustion efficiency** $\eta_c = 1 - \frac{\text{exhaust chemical energy as CO, H}_2, \text{HC, soot}}{\text{chemical energy in inducted fuel}}$
2. **Heat loss, finite combustion time, actual valve timing**

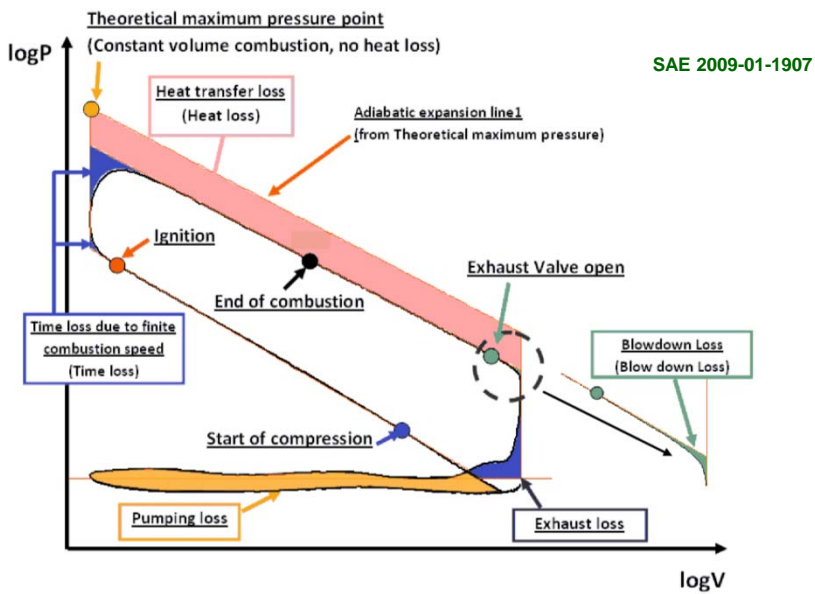


SI engine:
 H₂ and CO ~ 1 to 2% of fuel energy
 HC ~ 1% of fuel energy
 $\eta_c \sim 97-98\%$
 Diesel engine
 Very little unburned gas
 $\eta_c \sim 99\%$

Fig. 5-18
 Pressure-volume diagram for actual SI engine compared with that for equivalent fuel-air cycle; $r_c = 11$.

© McGraw-Hill Education. All rights reserved. This content is excluded from our Creative Commons license. For more information, see <https://ocw.mit.edu/help/faq-fair-use>.

Deconstruction of cycle losses



© Society of Automotive Engineers. All rights reserved. This content is excluded from our Creative Commons license. For more information, see <https://ocw.mit.edu/help/faq-fair-use>.

MIT OpenCourseWare
<https://ocw.mit.edu>

2.61 Internal Combustion Engines
Spring 2017

For information about citing these materials or our Terms of Use, visit: <https://ocw.mit.edu/terms>.